



Which of the following is a correct set of quantum numbers
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- A. $n = 5, l = 3, m_l = +1$
- B. $n = 5, l = 2, m_l = +3$
- C. $n = 4, l = 3, m_l = 0$
- D. $n = 4, l = 2, m_l = +1$

Show Answer...

Correct Answer: A. $n = 5, l = 3, m_l = +1$