



Which of the following statements is incorrect?

Which of the following statements is incorrect?

- A. dispersion force is the weakest type of intermolecular interactions
- B. the strong intermolecular attractions in H₂O result from hydrogen bonding
- C. boiling point of H₂S is less than H₂O
- D. boiling point of non-polar substances tends to decrease with increasing molecular weight

Show Answer...

Correct Answer: D. boiling point of non-polar substances tends to decrease with increasing molecular weight